The Rovibrational Intensities of the $2V_3$ Band of $^{12}C^{16}O^{18}O$ at 4639 cm⁻¹

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INTENSITIES OF 12C16O18O 2v3 BAND

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INTRODUCTION

Recent studies involving intensity measurements of weak CO₂ bands in the near-infrared¹⁻² were undertaken primarily to provide improved absorption parameters to model the night-side emission spectrum of Venus. Measurement of the intensity of some of these bands also provides the basis for improving the theoretical computation of many weaker CO₂ bands that cannot be measured. The hot, extensive atmosphere of Venus is primarily CO₂, so many bands that are weaker than are listed in the HITRAN compilation³ are needed for complete modeling of Venus' atmosphere. For most CO₂ bands, intensity measurements need only be made for the major isotope ¹²C¹⁶O₂, which could provide the basis for calculating the intensities of the corresponding bands of the minor isotopes. However, some transitions are forbidden in the symmetric isotopomers ¹²C¹⁶O₂ and ¹³C¹⁶O₂; absorption bands are only observed for these transitions in the asymmetric isotopomers. An important band of this type near the 2.2 µm emission window in Venus' nightside spectrum is the 2v₃ band of ¹²C¹⁶O¹⁸O at 4639 cm⁻¹.

The strength of this band currently used in the HITRAN compilation³ was obtained by Abubakar and Shaw⁴ using the technique of nonlinear least-squares whole-band analysis. Their two spectra had resolution of 0.06 cm⁻¹ and signal/noise of 130 and 260. With several spectra available in this wavenumber region that are about 5 times better in both resolution and signal/noise, we expected that measurements of individual lines could improve the intensity parameters of this band. We therefore selected the four spectra on

Line intensities were determined using non-linear leastsquares fitting of the spectra⁷. Line broadening coefficients are required for these intensity determinations, but we did not adopt the HITRAN³ CO₂ self-broadening values. The HITRAN values are based on careful measurements of several 12C16O2 bands, and are assumed applicable to all bands, regardless of vibrational transition or CO₂ isotopomer. We decided to investigate this assumption for this 2v₃ band since for ¹²C¹⁶O¹⁸O in normal isotopic abundance, broadening is mostly "foreign" broadening by 12C16O2. The latter may be somewhat different from the HITRAN self-broadening values. We therefore included the determination of the "foreign" broadening parameters. γ_f , in the measurements of the best isolated lines on spectrum No. 4; this spectrum was selected because the higher pressure spectrum with sufficiently deep lines should produce the more reliable broadening measurements. These values were then plotted against lml and smoothed; the smoothed values were adopted for intensity fits for all 4 spectra. These smoothed values are presented in the last column of Table 2. They are somewhat smaller (on the order of 3%) than HITRAN self-broadening coefficients at low-J values, but were indistinguishable for J>30.

We then determined γ_f parameters in isolated P-branch lines on spectrum No. 2, obtained at 45 torr, to check the reproducibility of our results from spectrum No. 4. The measurements from spectrum No. 2 were in very close agreement with HITRAN values, although we regard them as less reliable than the γ_f parameters from spectrum No. 4. The uncertainties in the γ_f parameters limit the absolute

DATA ANALYSIS

The measured line intensities in the units of cm⁻¹/(cm-atm) for each spectrum were converted to the standard temperature, 296 K, via the relation

$$\frac{S(T)}{S(T')} = \frac{Q_{vr}(T')}{Q_{vr}(T)} \left(\frac{T'}{T}\right) exp\left[\frac{-hcE''}{k} \left(\frac{1}{T} - \frac{1}{T'}\right)\right]. \tag{1}$$

The rovibrational partition function values, Q_{vr} , for $^{12}C^{16}O^{18}O$ were obtained from Gray and Young⁹ and the lower state rotational energy values E" (cm⁻¹) were adopted from the HITRAN³ tabulation. The intensity values were converted to units of cm⁻¹/(molecule/cm²) by dividing by 2.68676 x10¹⁹ (273.15/296). The rotationless transition moment squared and Herman-Wallis intensity parameters were obtained from the measured line intensities standardized to 296 K via the theoretical expression for the individual line intensities,

$$S_{j^*} = \{8\pi^3 10^{-36}/[3hcQ_{vr}(T)]\} \{\sigma \exp(-hcE''/kT)\} L_{j^*} |R_{vib}|^2 F(m), \qquad (2)$$

where the line intensity is in units of cm⁻¹/(molecule/cm²). The square of the rotationless transition moment $|R_{vib}|^2$ has units of Debye², J" is the lower state rotational quantum number, and σ (cm⁻¹) is the line position. The Hönl-London line strength factors $L_{J''}$ are equal to J" and (J"+1) respectively, for the P and R branches of the $2\nu_3$ band. T is the Kelvin temperature and k, h and c have their usual meanings.

parameters obtained from fitting Eq. 5 above are given in Table 3. Fig. 2 shows the plot of square root of reduced intensities versus m for each of the four spectra.

The rotationless band strength S_{vib}^0 , was evaluated at 296 K from the value of the rotationless transition moment:

$$S_{vib}^{0}(T) = \{8\pi^{3}10^{-36}/[3hcQ_{v}(T)]\}\sigma_{o}f |R_{vib}|^{2}$$
 (7)

using the value of the vibrational partition function, $Q_v(296K)=1.08724$, from Gray and Young⁹, and the vibrational band origin, $\sigma_o=4639.501$ cm⁻¹. This definition of S_{vib}^0 , which includes the normal isotopic abundance factor f, follows Rothman et al³ to be consistent with HITRAN usage. The result is included in Table 3.

DISCUSSION

The rotationless band strength we have obtained for the $2v_3$ band of $^{12}C^{16}O^{18}O$ in natural CO_2 is 16% larger than previously determined by Abubakar and Shaw⁴ who used the technique of nonlinear least-squares whole-band analysis. The Herman-Wallis parameter that we determined in Table 3 is very small and could not be determined with statistical significance. Therefore, we assumed F(m)=1 in the calculation of S_{cal} . The values of S_{cal} are given in Table 2. The precision of our intensity measurements, which on average is better than 1% as seen from column 4 of Table 2, takes account of uncertainties in the measurements of temperature and gas pressure in the cell. The relative dispersion of the intensity values as observed from column 6 of Table 2 is less than 1.5% on average.

on γ_f for these spectra with CO₂ pressures of 45 to 65 torr. Therefore, the likely systematic errors in the line intensities, and thus the band intensity, is less than $\pm 3\%$ from this main source of systematic uncertainties. Then, since $|R_{vib}|$ is proportional to $(S_{vib}^0)^{1/2}$, the likely systematic error associated with $|R_{vib}|$ is less than $\pm 1.5\%$.

Our tabulated line positions are the unweighted averages of the values measured and calibrated on our four spectra, which were obtained with CO₂ pressures from 45 to 65 torr; the average pressure is about 54 torr. These positions are only about 0.0005 cm⁻¹ smaller than the HITRAN³ zero-pressure line positions. The HITRAN line positions were calculated from J' upper level rotational constants determined from measurements in the 2v₃-v₃ band at 2307.4 cm⁻¹ which HITRAN attributed to M. P. Esplin in 1987. If our line positions and the HITRAN vlaues are both correct, the differences can be attributed to pressure shifts of about -0.007 cm⁻¹/atm. shift values¹¹ have complicated J' dependencies, but to first order the absolute value of pressure shifts are expected to depend linearly The HITRAN pressure shift values for the V₃ on wavenumber. fundamental band of ${}^{12}C^{16}O_2$ at 2349 cm⁻¹ are typically about -0.003 cm⁻¹/atm. Since our 2v₃ band of ¹²C¹⁶O¹⁸O is nearly a factor of 2 higher in frequency at 4639 cm⁻¹, a typical pressure shift for this band of -0.007 cm⁻¹/atm is very reasonable. Our line position measurements have therefore confirmed the HITRAN positions to well within their tabulated uncertainties of 0.001 cm⁻¹.

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Table 1

Spectrum Number	Path Length (m)	Pressure (Torr)	Temperature (K)	Resolution (cm ⁻¹)
1	97.04	65.0	297.5	0.012
2	409.82	45.0	297.8	0.012
3	145.16	45.0	298.0	0.012
4	193.28	59.7	298.0	0.012

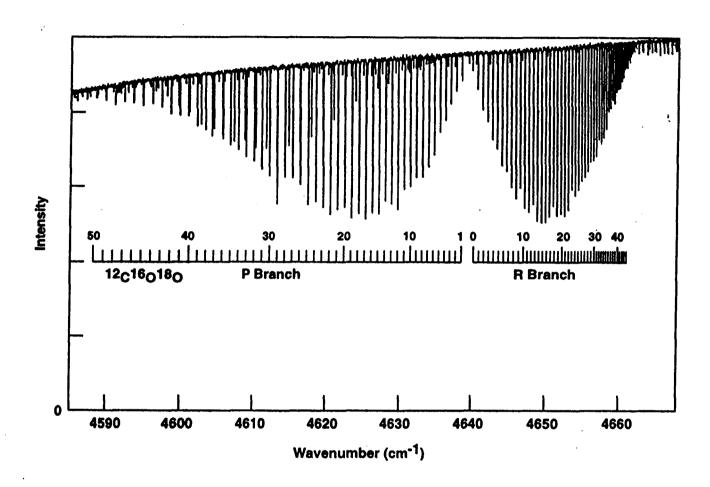


Fig. I.



TABLE 3

Parameter	
IR _{vib} l	$(1.4342 \pm 0.0007) \times 10^{-3} D$
aA ₁ bSo th	$(3.5 \pm 2.0) \times 10^{-5}$
^b S ^o vib	$(0.14416 \pm 0.00014) \times 10^{-22}$

^aThe A₁ Herman-Wallis parameter is dimensionless.

^bThe units of S⁰_{vib} are cm⁻¹/(molecule/cm²) of total CO₂ at 296 K.